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## Key indicators

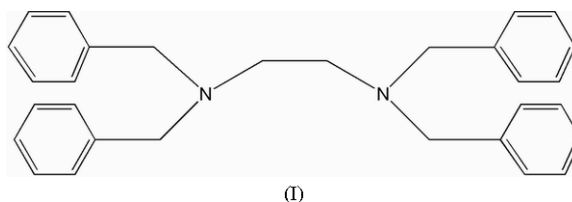
Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$   
 $R$  factor = 0.045  
 $wR$  factor = 0.181  
Data-to-parameter ratio = 14.8For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.*N,N,N',N'*-TetrabenzylethylenediamineThe title compound,  $\text{C}_{30}\text{H}_{32}\text{N}_2$ , was synthesized from ethylenediamine and benzyl chloride. The molecule occupies a special position on a crystallographic inversion centre.

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## Comment

Compounds incorporating the 1,2-diamine functionality are currently the topic of studies conducted in several fields (Lucet *et al.*, 1998). In our laboratory, we have recently investigated a new catalyst used in asymmetric synthesis. This compound, *N,N,N',N'*-tetrabenzylethylenediamine (TBEN), (I), was synthesized by the reaction of ethylenediamine and benzyl chloride. The structure determination of the title compound was undertaken as part of our studies on this catalyst.

The molecular structure of the title compound is illustrated in Fig. 1. The molecule occupies a special position on a crys-

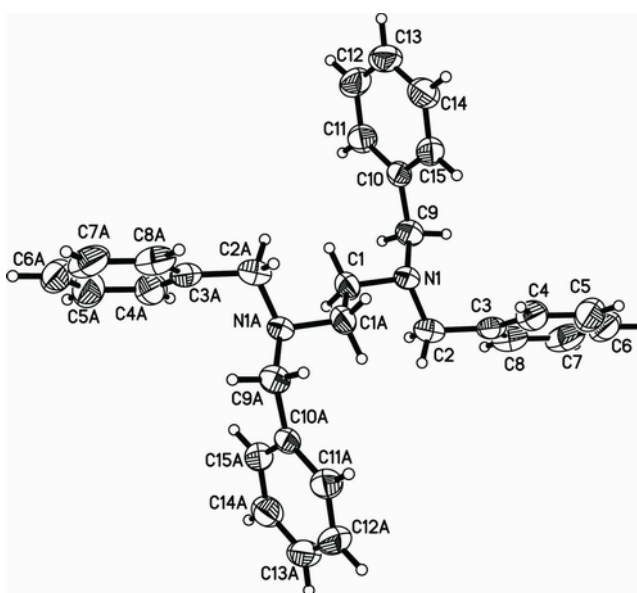


Figure 1

View of the title molecule, showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 35% probability level. The suffix A denotes an atom related by the symmetry operator  $-x, 2-y, -z$ .

tallographic inversion centre. Because of steric hindrance between the vicinal benzene rings, atom C2 has a slightly distorted tetrahedral geometry, with the N1–C2–C3 [113.40 (19)°] and N1–C9–C10 [114.50 (19)°] angles deviating significantly from the regular tetrahedral value (109.5°).

### Experimental

To an ethanol solution of ethylenediamine (1.00 mol), benzyl chloride (4.00 mol) was added dropwise with continuous stirring. The mixture was refluxed for 4 h. After cooling, the white hydrochloride of TBEN was filtered off, washed with ethanol and dried. The solid was then neutralized with 20% sodium hydroxide solution (400 ml) and the resulting oil was extracted; TBEN crystallized on standing. It was recrystallized from hot ethanol (m.p. 367–368 K). IR (KBr,  $\text{cm}^{-1}$ ): 3079, 3025, 2949, 2896, 1599, 1492, 1451, 1373, 1243, 1126, 1096, 1068, 1028, 976, 741, 697.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ,  $\delta$ , p.p.m.): 7.34–7.19 (*m*, 20H), 2(*s*, 8H), 4.81(*s*, 4H). Analysis calculated for  $\text{C}_{30}\text{H}_{32}\text{N}_2$ : C 85.70, H 7.60, N 6.65%; found: C 85.85, H 7.6, N 6.45%.

#### Crystal data

$\text{C}_{30}\text{H}_{32}\text{N}_2$	$Z = 1$
$M_r = 420.58$	$D_x = 1.135 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 5.8718 (19) \text{ \AA}$	Cell parameters from 960 reflections
$b = 10.202 (3) \text{ \AA}$	$\theta = 2.6\text{--}22.6^\circ$
$c = 10.629 (3) \text{ \AA}$	$\mu = 0.07 \text{ mm}^{-1}$
$\alpha = 97.621 (6)^\circ$	$T = 293 (2) \text{ K}$
$\beta = 101.621 (6)^\circ$	Block, colourless
$\gamma = 93.951 (6)^\circ$	$0.20 \times 0.18 \times 0.14 \text{ mm}$
$V = 615.2 (3) \text{ \AA}^3$	

#### Data collection

Bruker SMART CCD area-detector diffractometer	2166 independent reflections
$\varphi$ and $\omega$ scans	1283 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Bruker, 1997)	$R_{\text{int}} = 0.028$
$T_{\text{min}} = 0.926$ , $T_{\text{max}} = 0.991$	$\theta_{\text{max}} = 25.0^\circ$
3224 measured reflections	$h = -6 \rightarrow 6$
	$k = -12 \rightarrow 11$
	$l = -12 \rightarrow 12$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0994P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.181$	$(\Delta/\sigma)_{\text{max}} = 0.003$
$S = 1.06$	$\Delta\rho_{\text{max}} = 0.13 \text{ e \AA}^{-3}$
2166 reflections	$\Delta\rho_{\text{min}} = -0.13 \text{ e \AA}^{-3}$
146 parameters	Extinction correction: <i>SHELXL97</i>
H-atom parameters constrained	Extinction coefficient: 0.053 (15)

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

N1–C1	1.460 (3)	N1–C2	1.467 (3)
N1–C9	1.462 (3)		
C1–N1–C9	111.28 (17)	N1–C2–C3	113.40 (19)
C1–N1–C2	111.28 (17)	N1–C9–C10	114.15 (19)
C9–N1–C2	109.35 (19)		
C9–N1–C1–C1 <sup>i</sup>	–162.0 (3)	N1–C2–C3–C8	–126.5 (3)
C2–N1–C1–C1 <sup>i</sup>	75.8 (3)	C1–N1–C9–C10	68.8 (2)
C1–N1–C2–C3	–169.0 (2)	C2–N1–C9–C10	–167.90 (19)
C9–N1–C2–C3	67.7 (3)	N1–C9–C10–C15	52.2 (3)
N1–C2–C3–C4	55.4 (3)	N1–C9–C10–C11	–131.8 (2)

Symmetry code: (i)  $-x, 2 - y, -z$ .

All H atoms were positioned geometrically and refined as riding (C–H = 0.93–0.97  $\text{\AA}$ ) on the attached C atom. For both CH and  $\text{CH}_2$  groups, the  $U_{\text{iso}}(\text{H})$  values were set equal to  $1.2U_{\text{eq}}(\text{carrier atom})$ .

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

### References

- Bruker (1997). *SMART, SAINTE and SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Lucet, D., LeGall, T. & Mioskowski, C. (1998). *Angew. Chem. Int. Ed.* **37**, 2581–2627.
- Sheldrick, G. M. (1997). *SHELXS97 and SHELXL97*. University of Göttingen, Germany.